

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Flavor West Manufacturing, LLC.

Chemwatch Hazard Alert Code: 2

Issue Date: 04/10/2015 Print Date: 05/14/2015 Initial Date: 04/11/2015 L.GHS.USA.EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

### **Product Identifier**

Version No: 1.1

Product name	FW-CCN N&A Candy Corn Flavor
Synonyms	Not Available
Other means of identification	Not Available

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified	Use according to manufacturer's directions.
uses	

### Details of the manufacturer/importer

Registered company name	Flavor West Manufacturing, LLC.
Address	29400 Hunco Way, Lake Elsinore CA 92530 United States
Telephone	(951) 893-5120
Fax	(714) 276-1621
Website	www.FlavorWest.com
Email	Flavor@FlavorWest.com

### **Emergency telephone number**

Association / Organisation	Chemwatch
Emergency telephone numbers	see below
Other emergency telephone numbers	see below

### CHEMWATCH EMERGENCY RESPONSE

Primary Number	Alternative Number 1	Alternative Number 2
877 715 9305	+612 9186 1132	Not Available

Once connected and if the message is not in your prefered language then please dial 01

Una vez conectado y si el mensaje no está en su idioma preferido, por favor marque 02

### **SECTION 2 HAZARDS IDENTIFICATION**

Classification of the substance or mixture



GHS Classification Skin Corrosion/Irritation Category 2, Skin Sensitizer Category 1, Eye Irritation Category 2A

#### Label elements

GHS label elements	
SIGNAL WORD	WARNING

#### Hazard statement(s)

H315	Causes skin irritation
H317	May cause an allergic skin reaction
H319	Causes serious eye irritation

### Precautionary statement(s) Prevention

	P280	P280 Wear protective gloves/protective clothing/eye protection/face protection.	
P261 Avoid breathing dust/fume/gas/mist/vapours/spray.		Avoid breathing dust/fume/gas/mist/vapours/spray.	
	P272	Contaminated work clothing should not be allowed out of the workplace.	

#### Precautionary statement(s) Response

P362	Take off contaminated clothing.	
P363	Wash contaminated clothing before reuse.	
P302+P352	IF ON SKIN: Wash with plenty of water and soap	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.	
P337+P313	If eye irritation persists: Get medical advice/attention.	

# Precautionary statement(s) Storage

#### Precautionary statement(s) Disposal

P501

Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

#### SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### Substances

See section below for composition of Mixtures

### **Mixtures**

CAS No	%[weight]	Name
57-55-6	85-90	propylene glycol
121-32-4	5-10	ethyl vanillin
121-33-5	1-5	vanillin
119-84-6	1-5	dihydrocoumarin

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

### **SECTION 4 FIRST AID MEASURES**

Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Wash out immediately with fresh running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
Skin Contact	<ul> <li>If skin contact occurs:</li> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available).</li> <li>Seek medical attention in event of irritation.</li> </ul>
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor.</li> </ul>
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> </ul>

### Description of first aid measures

### Indication of any immediate medical attention and special treatment needed

- Polyethylene glycols are generally poorly absorbed orally and are mostly unchanged by the kidney.
- Dermal absorption can occur across damaged skin (e.g. through burns) leading to increased osmolality, anion gap metabolic acidosis, elevated calcium, low ionised calcium, CNS depression and renal failure.
- Treatment consists of supportive care.

[Ellenhorn and Barceloux: Medical Toxicology]

Propylene glycol is primarily a CNS depressant in large doses and may cause hypoglycaemia, lactic acidosis and seizures.

- The usual measures are supportive care and decontamination (Ipecac/ lavage/ activated charcoal/ cathartics), within 2 hours of exposure should suffice.
- Check the anion gap, arterial pH, renal function and glucose levels.

Ellenhorn and Barceloux: Medical Toxicology

#### SECTION 5 FIREFIGHTING MEASURES

### Extinguishing media

<ul> <li>Alcohol stable foam.</li> <li>Dry chemical powder.</li> <li>BCF (where regulations permit).</li> <li>Carbon dioxide.</li> </ul>

#### Special hazards arising from the substrate or mixture

Fire Incompatibility	• Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may
The meonpationity	result

#### Advice for firefighters

Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Use water delivered as a fine spray to control fire and cool adjacent area.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Combustible.</li> <li>Slight fire hazard when exposed to heat or flame.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>On combustion, may emit toxic fumes of carbon monoxide (CO).</li> </ul>

### SECTION 6 ACCIDENTAL RELEASE MEASURES

Minor Spills	<ul> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> </ul>								
	Chemical Class: alcor For release onto land	۰,		sorbents listed in o	rder o	of p	priority.		
	SORBENT TYPE	RANK		APPLICATION			COLLECTION		LIMITATIONS
	LAND SPILL - SMALL	-							
	cross-linked polymer	- particulat	te		1		shovel	shovel	R, W, SS
	cross-linked polymer	- pillow			1		throw	pitchfork	R, DGC, RT
	sorbent clay - particu	late			2		shovel	shovel	R,I, P
	wood fiber - pillow				3		throw	pitchfork	R, P, DGC, RT
	treated wood fiber - p	illow			3		throw	pitchfork	DGC, RT
	foamed glass - pillow	,			4		throw	pichfork	R, P, DGC, RT
	LAND SPILL - MEDIUM								
	cross-linked polymer	cross-linked polymer - particulate			1	b	lower	skiploader	R,W, SS
Major Spills	polypropylene - particulate			2	b	lower	skiploader	W, SS, DGC	
	sorbent clay - particulate			2	b	lower	skiploader	R, I, W, P, DGC	
	polypropylene - mat				3	th	nrow	skiploader	DGC, RT
	expanded mineral - p	expanded mineral - particulate			3	b	lower	skiploader	R, I, W, P, DGC
	polyurethane - mat				4	tŀ	nrow	skiploader	DGC, RT
	Legend DGC: Not effective where ground cover is dense R; Not reusable I: Not incinerable P: Effectiveness reduced when rainy RT:Not effective where terrain is rugged SS: Not for use within environmentally sensitive sites W: Effectiveness reduced when windy Reference: Sorbents for Liquid Hazardous Substance Cleanup and Control; R.W Melvold et al: Pollution Technology Review No. 150: Noyes Data Corporation 1988 Moderate hazard. • Clear area of personnel and move upwind.								
	<ul> <li>Alert Fire Brigade a</li> </ul>	and tell them	n loc	ation and nature of					

# SECTION 7 HANDLING AND STORAGE

### Precautions for safe handling

Safe handling	<ul> <li>DO NOT allow clothing wet with material to stay in contact with skin</li> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> </ul>
Other information	<ul> <li>Material is hygroscopic, i.e. absorbs moisture from the air. Keep containers well sealed in storage.</li> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> </ul>

# Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Metal can or drum</li> <li>Packaging as recommended by manufacturer.</li> </ul>
	<ul> <li>Check all containers are clearly labelled and free from leaks.</li> </ul>

	<ul> <li>Glycols and their ethers undergo violent decomposition in contact with 70% perchloric acid. This seems likely to involve formation of the glycol perchlorate esters (after scission of ethers) which are explosive, those of ethylene glycol and 3-chloro-1,2-propanediol being more powerful than glyceryl nitrate, and the former so sensitive that it explodes on addition of water.</li> <li>Alcohols</li> </ul>
Storage	are incompatible with strong acids, acid chlorides, acid anhydrides, oxidising and reducing agents.
incompatibility	reacts, possibly violently, with alkaline metals and alkaline earth metals to produce hydrogen
	<ul> <li>react with strong acids, strong caustics, aliphatic amines, isocyanates, acetaldehyde, benzoyl peroxide, chromic acid, chromium oxide, dialkylzincs, dichlorine oxide, ethylene oxide, hypochlorous acid, isopropyl chlorocarbonate, lithium tetrahydroaluminate, nitrogen dioxide, pentafluoroguanidine, phosphorus halides, phosphorus pentasulfide, tangerine oil, triethylaluminium, triisobutylaluminium</li> <li>should not be heated above 49 deg.</li> </ul>

#### PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

### SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

### **Control parameters**

#### OCCUPATIONAL EXPOSURE LIMITS (OEL)

# INGREDIENT DATA

Not Available

#### EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
propylene glycol	Propylene glycol; (1,2-Propanediol)	30 mg/m3	1300 mg/m3	7900 mg/m3
vanillin	Vanilin	10 mg/m3	10 mg/m3	310 mg/m3

Ingredient	Original IDLH	Revised IDLH
propylene glycol	Not Available	Not Available
ethyl vanillin	Not Available	Not Available
vanillin	Not Available	Not Available
dihydrocoumarin	Not Available	Not Available

### MATERIAL DATA

for propylene glycol:

Saturated vapour concentration @ 20 deg C.= 65.8 ppm, 204.6 mg/m3; i.e higher concentrations can only occur as aerosols or at higher temperatures. Odour Threshold: Practically odourless.

A small number of individuals show skin irritation or sensitisation from repeated or prolonged exposure to propylene glycol. A workplace environmental exposure limit (WEEL) has been established by AIHA and is thought to be protective against systemic effects.

#### Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.
Personal protection	
Eye and face protection	<ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task.</li> </ul>
Skin protection	See Hand protection below
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber</li> <li>NOTE:</li> <li>The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and</li> </ul>

	<ul> <li>other protective equipment, to avoid all possible skin contact.</li> <li>Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.</li> </ul>
Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>P.V.C. apron.</li> <li>Barrier cream.</li> </ul>
Thermal hazards	Not Available

### Recommended material(s)

### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the: "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

FW-CCN N&A Candy Corn Flavor

Material	СРІ
BUTYL	С
NATURAL RUBBER	С
NEOPRENE	С
PE/EVAL/PE	С
PVA	С
VITON	С

\* CPI - Chemwatch Performance Index

Appearance

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion C: Poor to Dangerous Choice for other than short term immersion **NOTE**: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

#### Information on basic physical and chemical properties

Clear colorless

#### Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator	
up to 10 x ES	A-AUS P2	-	A-PAPR-AUS / Class 1 P2	
up to 50 x ES	-	A-AUS / Class 1 P2	-	
up to 100 x ES	-	A-2 P2	A-PAPR-2 P2 ^	

#### ^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

Physical state	Liquid	Relative density (Water = 1)	1.06
Odour	Characteristic	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	> 93.00	Taste	Candy corn
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available

Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

# SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

# SECTION 11 TOXICOLOGICAL INFORMATION

# Information on toxicological effects

Inhaled	The material is not thought to produce respiratory irritation (as classified by EC Directives using animal models). Nevertheless inhalation of vapours, fumes or aerosols, especially for prolonged periods, may produce respiratory discomfor and occasionally, distress. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by narcosis, reduced alertness, loss of reflexes, lack of coordination and vertigo.		
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual. Ingestion of propylene glycol produced reversible central nervous system depression in humans following ingestion of 60 ml. Symptoms included increased heart-rate (tachycardia), excessive sweating (diaphoresis) and grand mal seizures in a 15 month child who ingested large doses (7.5 ml/day for 8 days) as an ingredient of vitamin preparation. Excessive repeated ingestions may cause hypoglycaemia (low levels of glucose in the blood stream) among susceptible individuals; this may result in muscular weakness, incoordination and mental confusion.		
Skin Contact	<ul> <li>Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions.</li> <li>The material may produce moderate skin irritation; limited evidence or practical experience suggests, that the material either:</li> <li>produces moderate inflammation of the skin in a substantial number of individuals following direct contact and/or</li> <li>produces significant, but moderate, inflammation when applied to the healthy intact skin of animals (for up to four hours), such inflammation being present twenty-four hours or more after the end of the exposure period.</li> <li>Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis.</li> </ul>		
Eye	Limited evidence exists, or practical experience suggests, that the material may cause eye irritation in a substantial number of individuals and/or is expected to produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur. Irritation of the eyes may produce a heavy secretion of tears (lachrymation).		
Chronic	of individuals at a greater frequency than would be expecte	sitive response in experimental animals. pational exposure may produce cumulative health effects pable of inducing a sensitisation reaction in a significant number	
FW-CCN N&A Candy Corn Flavor	TOXICITY Not Available	IRRITATION Not Available	
propylene glycol	TOXICITY Dermal (rabbit) LD50: >2000 mg/kg <sup>[1]</sup>	IRRITATION Eye (rabbit): 100 mg - mild	

	Oral (rat) LD50: 20000 mg/kgd <sup>[2]</sup>	Eye (rab	bit): 500 mg/24	h - mild
		Skin(hun	nan):104 mg/3d	Intermit Mod
		Skin(hun	nan):500 mg/7c	ays mild
	ΤΟΧΙΟΙΤΥ		IRRITATION	
	Dermal (rabbit) LD50: >7940 mg/kg* <sup>[2]</sup>	* [Monsanto]		
ethyl vanillin	Oral (rat) LD50: 1590 mg/kg* <sup>[2]</sup>		Eye (rabbit): 1	0/110.0 *
			Skin (rabbit): 0	.3/8.0 slight *
	TOXICITY			IRRITATION
vanillin	dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>			Not Available
	Oral (rat) LD50: 1400 mg/kg <sup>[1]</sup>			
	ΤΟΧΙΟΙΤΥ	IRRI	TATION	
dihydrocoumarin	Dermal (rabbit) LD50: >5000 mg/kg <sup>[2]</sup>	Skir	ı (g.pig): 1%/48	h - mod
	Oral (rat) LD50: 1460 mg/kgd <sup>[2]</sup>	Skir	(rabbit): 500 m	ng/24h - mod
		Outotances A. ( )		taling of function and the start of the start
Legend:	<ol> <li>Value obtained from Europe ECHA Registered Unless otherwise specified data extracted from I</li> </ol>		-	
	The acute oral toxicity of propylene glycol is ve damage in humans.			
ETHYL VANILLIN	Asthma-like symptoms may continue for mont to a non-allergenic condition known as reactive exposure to high levels of highly irritating com preceding respiratory disease, in a non-atopic minutes to hours of a documented exposure to of moderate to severe bronchial hyperreactivit inflammation, without eosinophilia, have also b	e airways dysfunction syn pound. Key criteria for th individual, with abrupt or o the irritant. A reversible by on methacholine challe	ndrome (RADS ne diagnosis of nset of persiste airflow pattern enge testing ar	) which can occur following RADS include the absence of nt asthma-like symptoms within , on spirometry, with the presence d the lack of minimal lymphocytic
VANILLIN	For certain benzyl derivatives: All members of this group (benzyl, benzoate ar directly to an oxygenated functional group (ald derivative. As a stable animal metabolite, ben: reaction pathways have been reported in both is a reflection their participation in these comm Miosis, somnolence, muscle weakness, coma uterus, cervix and vagina recorded.	lehyde or ester) that is hy zoic acid derivatives are aquatic and terrestrial sp non metabolic pathways.	vdrolysed and/o efficiently excr pecies. The sim	or oxidised to a benzoic acid eted primarily in the urine. These ilarity of their toxicologic properties
DIHYDROCOUMARIN	Somnolence recorded.			
FW-CCN N&A Candy Corn Flavor & DIHYDROCOUMARIN	The following information refers to contact alle Contact allergies quickly manifest themselves pathogenesis of contact eczema involves a ce allergic skin reactions, e.g. contact urticaria, ir	as contact eczema, mor ell-mediated (T lymphocy	e rarely as urtio tes) immune re	caria or Quincke's oedema. The eaction of the delayed type. Other
Acute Toxicity	0	Carcinogenic	ity 🛇	
Skin				
Irritation/Corrosion	*	Reproductiv	ity 🛇	
Serious Eye Damage/Irritation	*	STOT - Sing Exposu		
Respiratory or Skin sensitisation	<b>v</b>	STOT - Repeat Exposu		

Mutagenicity	S Aspiration Hazard S	
	<ul> <li>Legend: ✓ – Data required to make classification available</li> <li>X – Data available but does not fill the criteria for classification</li> <li>S – Data Not Available to make classification</li> </ul>	
CMR STATUS		
CARCINOGEN	dihydrocoumarin US Environmental Defense Scorecard Suspected Carcinogens CPDB	

# SECTION 12 ECOLOGICAL INFORMATION

### Toxicity

# NOT AVAILABLE

Ingredient	Endpoint	Test Duration	Effect	Value	Species	BCF
propylene glycol	Not Available					
ethyl vanillin	Not Available					
vanillin	Not Available					
dihydrocoumarin	Not Available					

Propylene glycol is known to exert high levels of biochemical oxygen demand (BOD) during degradation in surface waters. This process can adversely affect aquatic life by consuming oxygen needed by aquatic organisms for survival. Large quantities of dissolved oxygen (DO) in the water column are consumed when microbial populations decompose propylene glycol.

Sufficient dissolved oxygen levels in surface waters are critical for the survival of fish, macro-invertebrates, and other aquatic organisms.

### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
propylene glycol	LOW	LOW
ethyl vanillin	LOW	LOW
vanillin	LOW	LOW
dihydrocoumarin	LOW	LOW

### **Bioaccumulative potential**

Ingredient	Bioaccumulation
propylene glycol	LOW (BCF = 1)
ethyl vanillin	LOW (LogKOW = 1.58)
vanillin	LOW (LogKOW = 1.21)
dihydrocoumarin	LOW (LogKOW = 0.967)

### Mobility in soil

Ingredient	Mobility
propylene glycol	HIGH (KOC = 1)
ethyl vanillin	LOW (KOC = 70.92)
vanillin	LOW (KOC = 38.45)
dihydrocoumarin	LOW (KOC = 146.1)

# SECTION 13 DISPOSAL CONSIDERATIONS

#### Waste treatment methods

	<ul> <li>Containers may still present a chemical hazard/ danger when empty.</li> <li>Return to supplier for reuse/ recycling if possible.</li> </ul>
Product / Packaging	Otherwise:
disposal	• If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to
	store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
	Where possible retain label warnings and MSDS and observe all notices pertaining to the product.

### **SECTION 14 TRANSPORT INFORMATION**

#### Labels Required

Marine Pollutant

NO

#### Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

### Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

### Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

#### Transport in bulk according to Annex II of MARPOL 73 / 78 and the IBC code

Source	Ingredient	Pollution Category
IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk	dihydrocoumarin	X

# SECTION 15 REGULATORY INFORMATION

#### Safety, health and environmental regulations / legislation specific for the substance or mixture

propylene glycol(57-55-6) is found on the following regulatory lists	"US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)","US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values","US AIHA Workplace Environmental Exposure Levels (WEELs)","US Spacecraft Maximum Allowable Concentrations (SMACs) for Airborne Contaminants","US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
ethyl vanillin(121-32-4) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
vanillin(121-33-5) is found on the following regulatory lists	"US AIHA Workplace Environmental Exposure Levels (WEELs)", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
dihydrocoumarin(119-84-6) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	Y
Korea - KECI	Y
New Zealand - NZIoC	Y
Philippines - PICCS	Y
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

### **SECTION 16 OTHER INFORMATION**

### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

### www.chemwatch.net

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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